

AMINOMETHANESULFONIC ACID AND ITS N-ALKYLATED DERIVATIVES. SYNTHESIS, CHARACTERIZATION AND TOXICITY STUDIES

Ruslan Khoma ^{a,b}, Vyacheslav Baumer ^c, Petro Antonenko ^d, Anastasiia Snihach ^d,
Vladlena Godovan ^d, Alim Ennan ^a, Ruslan Dlubovskii ^a, Vladimir Gelmboldt ^d

^aPhysical-Chemical Institute for Environment and Human Protection of MES of Ukraine and NAS of Ukraine,
3, Preobrazhenska str., Odessa 65082, Ukraine

^bOdessa I.I. Mechnikov National University, 2, Dvoryanskaya str., Odessa 65000, Ukraine

^cInstitute of Single Crystals of the National Academy of Sciences of Ukraine, Lenin ave., 60, Kharkov 61001, Ukraine

^dOdessa National Medical University, Valikhovskiy lane, 2, Odessa 65082, Ukraine

*e-mail: rek@onu.edu.ua

Abstract. Aminomethanesulfonic acid (**1a**), and its N-methyl (**2a**), N-n-propyl (**3a**), N-tert-butyl (**4a**) derivatives have been synthesized (**3a** was not previously described) and characterized by means of X-ray diffraction analysis, IR spectroscopy, and mass spectrometry. The aqueous solutions of sodium salts (**1b–4b**) were prepared by reacting NaOH with the corresponding **1a–4a** acids in equimolar amounts. The toxicity of **1b–4b** has been studied on 132 mice. The aminomethanesulfonic acid derivatives belong to almost non-toxic compounds (V category of toxicity) following both intraperitoneal and oral administration in mice. According to the LD50 values, the compounds could be ranked in the following order: **3b** (2110 mg/kg) < **2b** (2925 mg/kg) < **4b** (3020 mg/kg) < **1b** (3470 mg/kg). Aminomethanesulfonic acid derivatives may be safely used in buffer solution as well as for further investigation as potential medicine agents.

Keywords: aminomethanesulfonic acid, synthesis, crystal structure, lethal toxicity.